

Production of FT-diesel

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Bioenergy 2020+

Vienna, University of Technology



Nationaler Workshop Biotreibstoffe
29th September 2016



Fulcrum Bioenergy Sierra Biofuels Plant

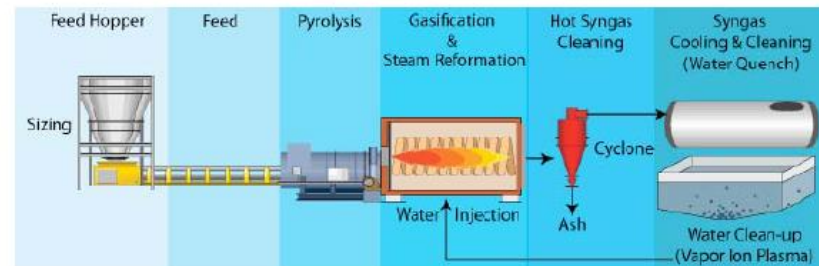
- Waste to FT fuels
- 200,000 t/y MSW
- Designed for 10 million gallons syncrude
- TRI gasifier
- Agreement with United Airlines
- Collaboration with Tesoro
- Startup expected late 2017





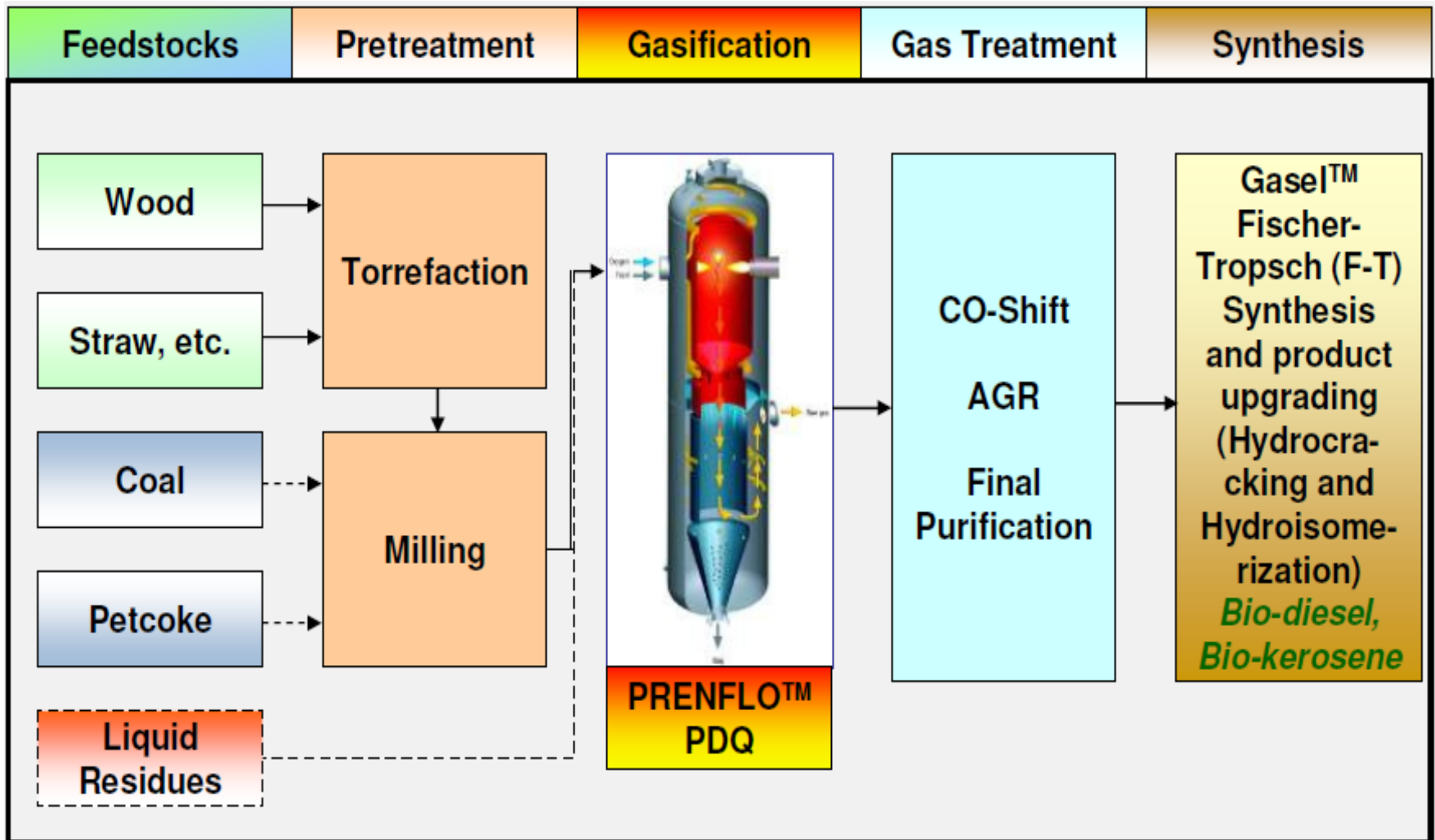
Red Rock Biofuels

- Biomass to FT fuels
- Lakeview, Oregon
- TCG Global gasifier
- Veolocys for FT
- Targeting jet fuel
- Target completion 2016
- \$182 million investment
- Recent partnership announced with FedEx
 - provide 3 million gal/y of biofuels
 - blended to 7 million gal/y
 - 2017 to 2024





BioTfuel, Frankreich



Synthetic Biofuels (FT- Route)

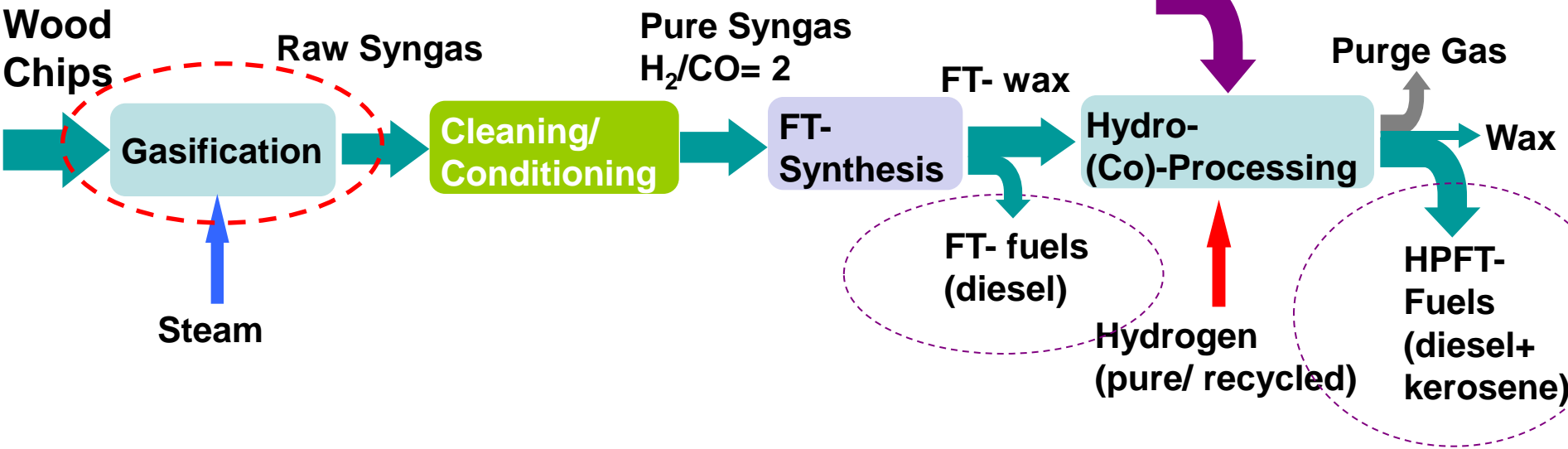


Cellulose
Hemicellulose
Lignin



i/n- paraffins
(hydrocarbons)

Fossil Products
(e.g. LGO, HGO, VGO)



Location	Usage / Product	Fuel / Product MW, MW	Start up	Supplier	Status
Güssing, AT	Gas engine	8.0 _{fuel} / 2.0 _{el}	2002	AE&E, Repotec	Operational
Oberwart, AT	Gas engine / ORC / H ₂	8.5 _{fuel} / 2.8 _{el}	2008	Ortner Anlagenbau	(Operational)
Villach, AT	Gas engine	15 _{fuel} / 3.7 _{el}	2010	Ortner Anlagenbau	On hold
Senden/Ulm, DE	Gas engine / ORC	14 _{fuel} / 5 _{el}	2011	Repotec	Operational
Burgeis, IT	Gas engine	2 _{fuel} / 0.5 _{el}	2012	Repotec, RevoGas	(Operational)
Göteborg, Sweden	BioSNG	32 _{fuel} /20 _{BioSNG}	2013	Repotec/ Valmet	Operational
California	R&D	1 MW _{fuel}	2013	GREG	Operational
Gaya, France	BioSNG R&D	0,5 MW _{fuel}	2016	Repotec	Commissioning
Thailand	Gas engine	4 _{fuel} / 1 _{el}	2016	GREG	Under construction

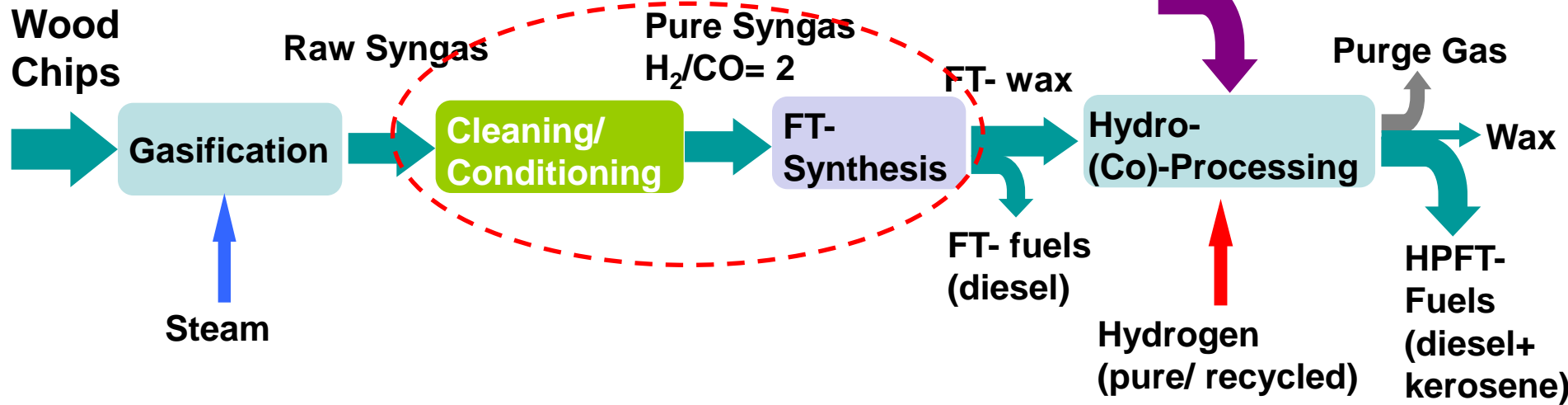
Synthetic Biofuels (FT- Route)

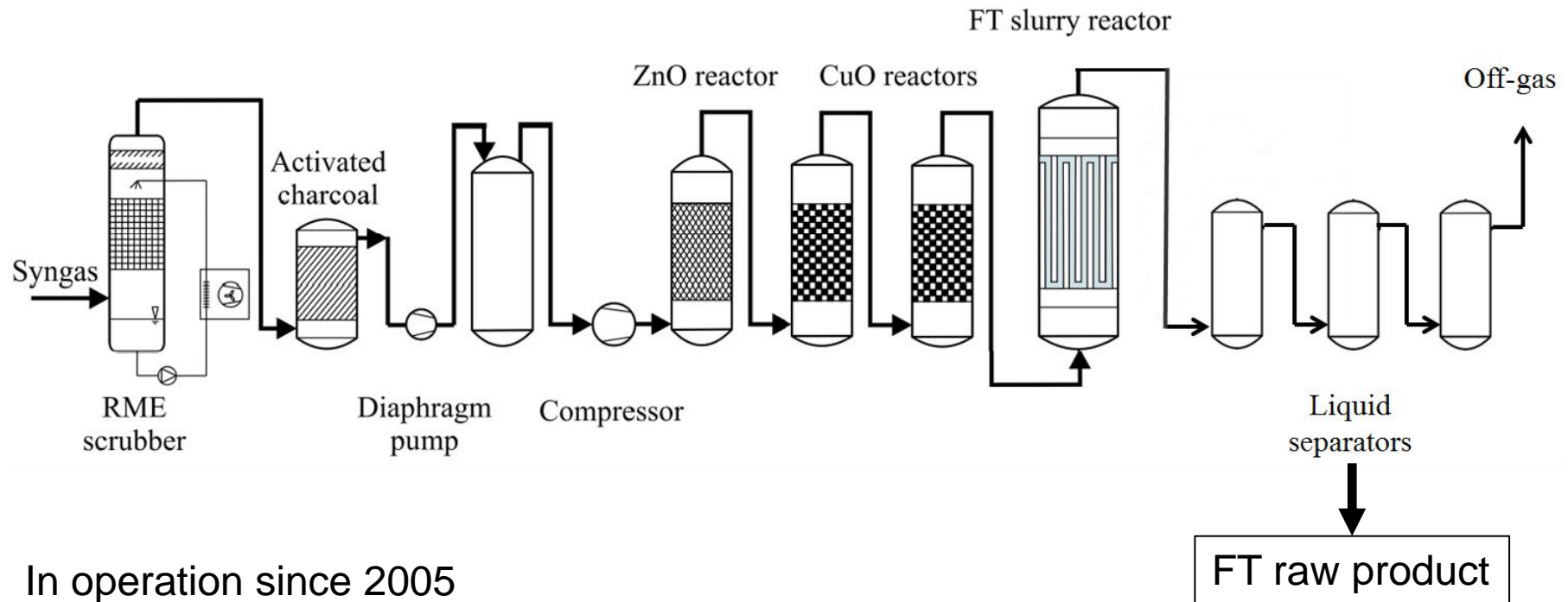


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In operation since 2005

5-10kg/day of FT raw product

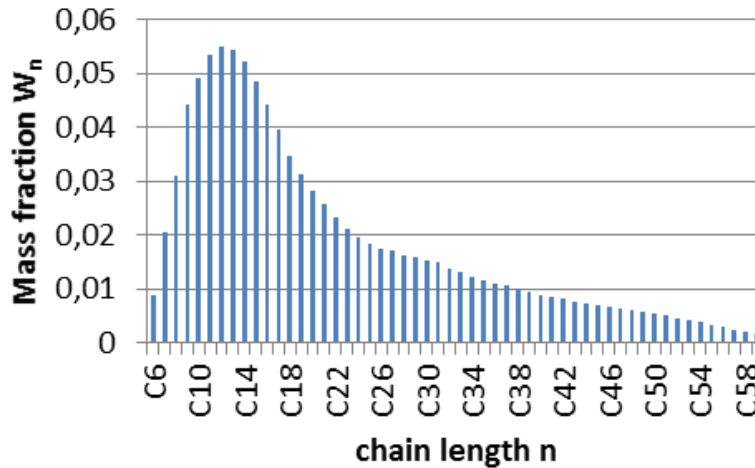
Slurry reactor, because of excellent heat transfer and easy scaling up

Gas treatment removes Sulphur to below 10ppb

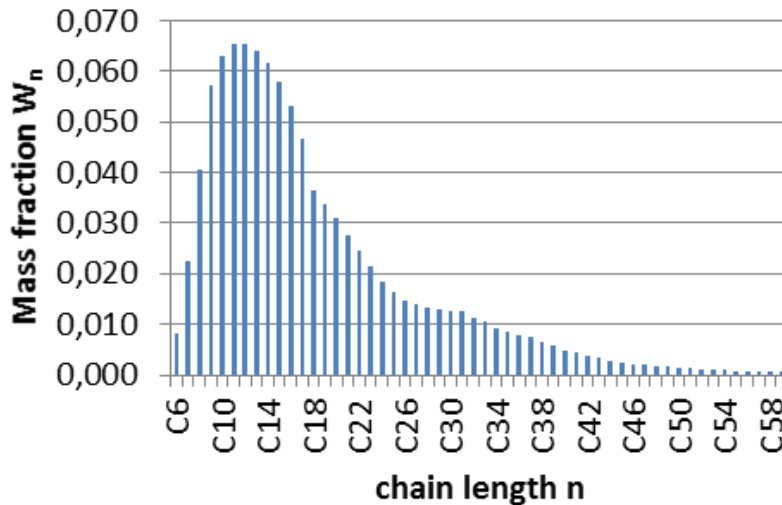
Cobalt and Iron- based catalyst were tested

Fully automatic

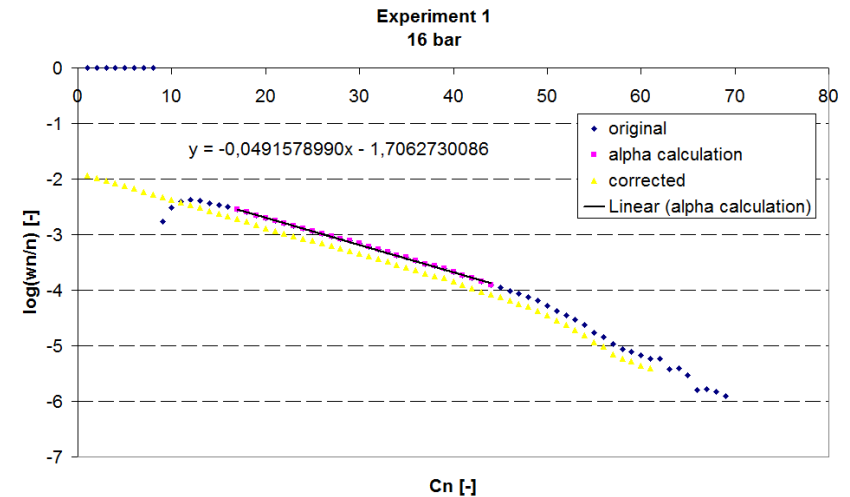
Product distribution 1000 h cat A



Product distribution 500 h catalyst D

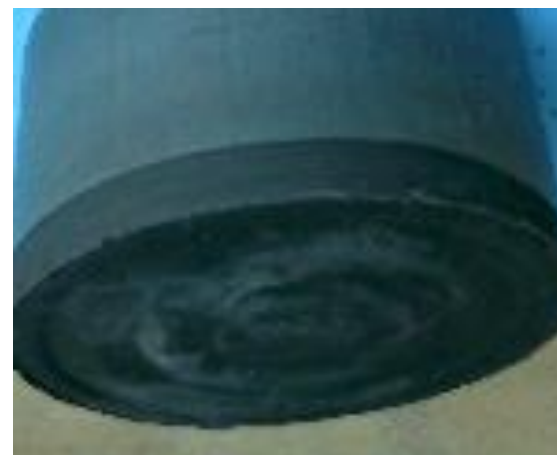
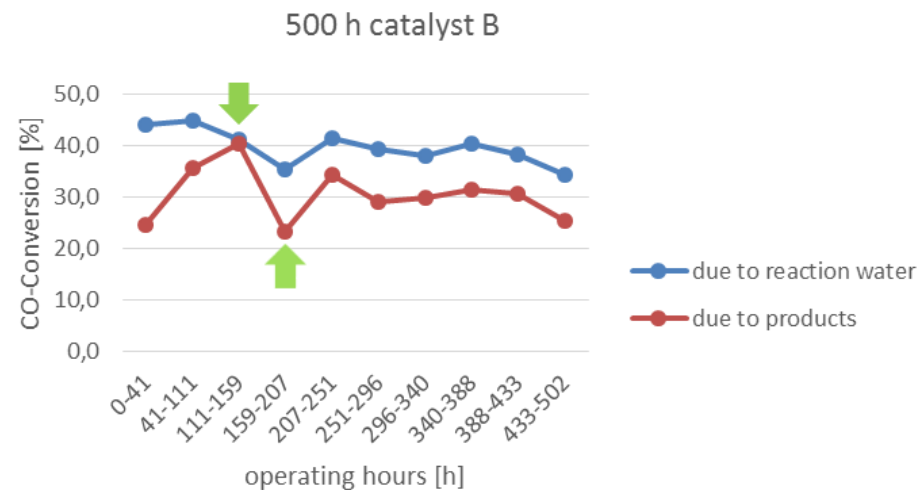
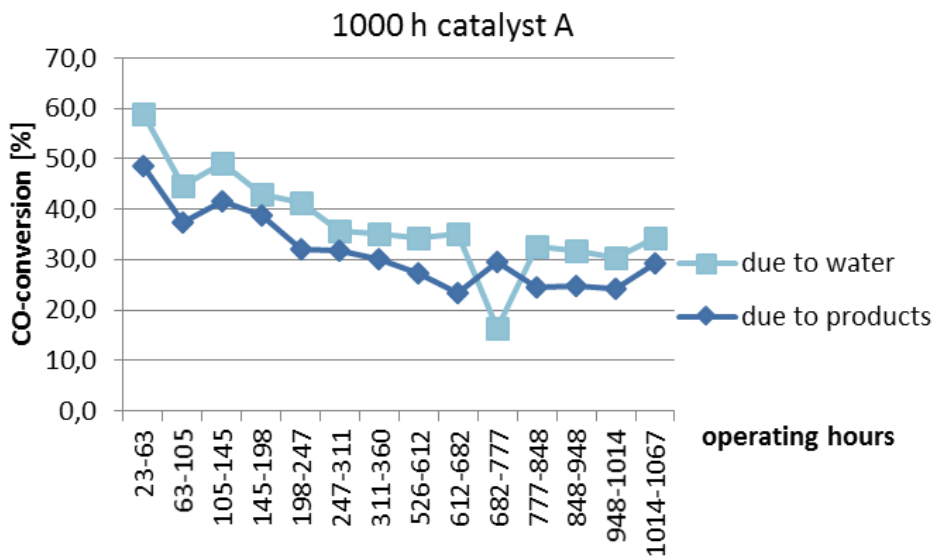


Experimental data for input in simulation of M&E balances



$$W_n = n(1 - \alpha)^2 \cdot \alpha^{n-1}$$

$$\log \frac{W_n}{n} = n \log(\alpha) + \log \frac{(1 - \alpha)^2}{\alpha}$$



**5 liters per day
in operation**

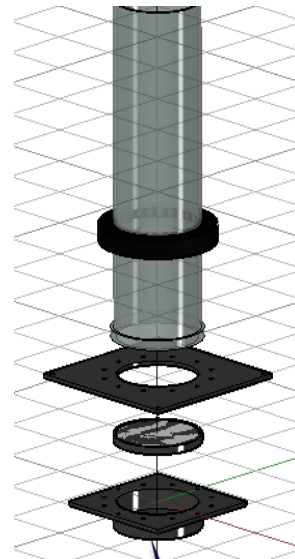


Smaller Cold Flow Model
(100 mm inner diameter Ø)

Hot Reactor
and Pilot Plant



**1 barrel per day
In commissioning**

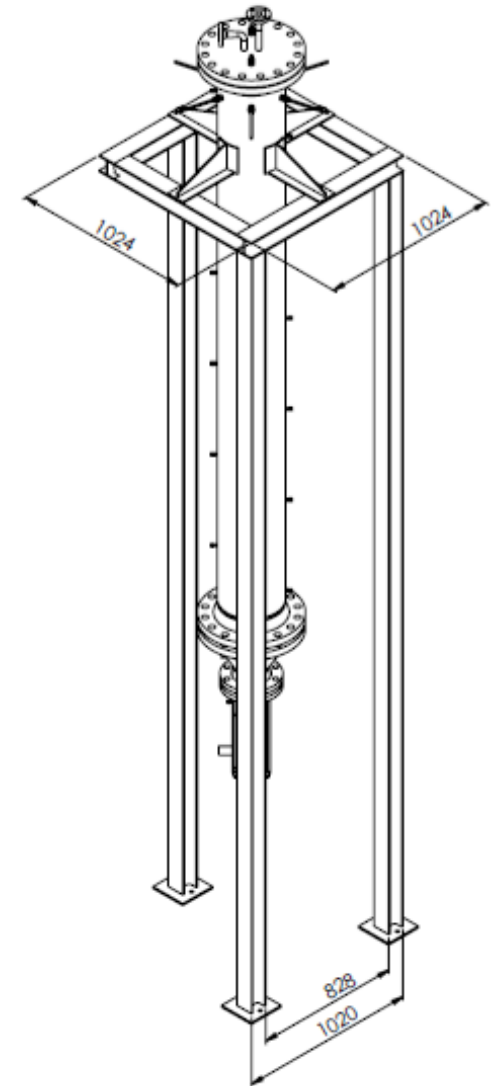
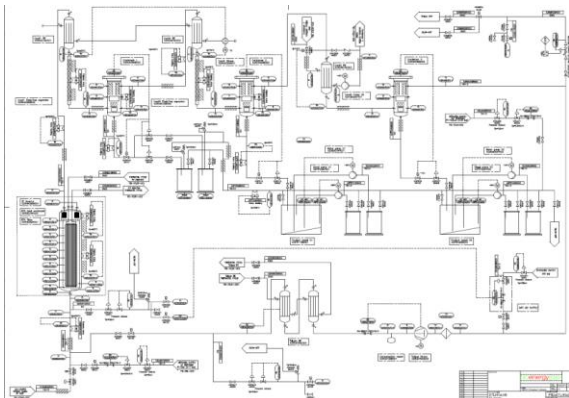


Design for the larger Cold
Flow Model (300 mm Ø)

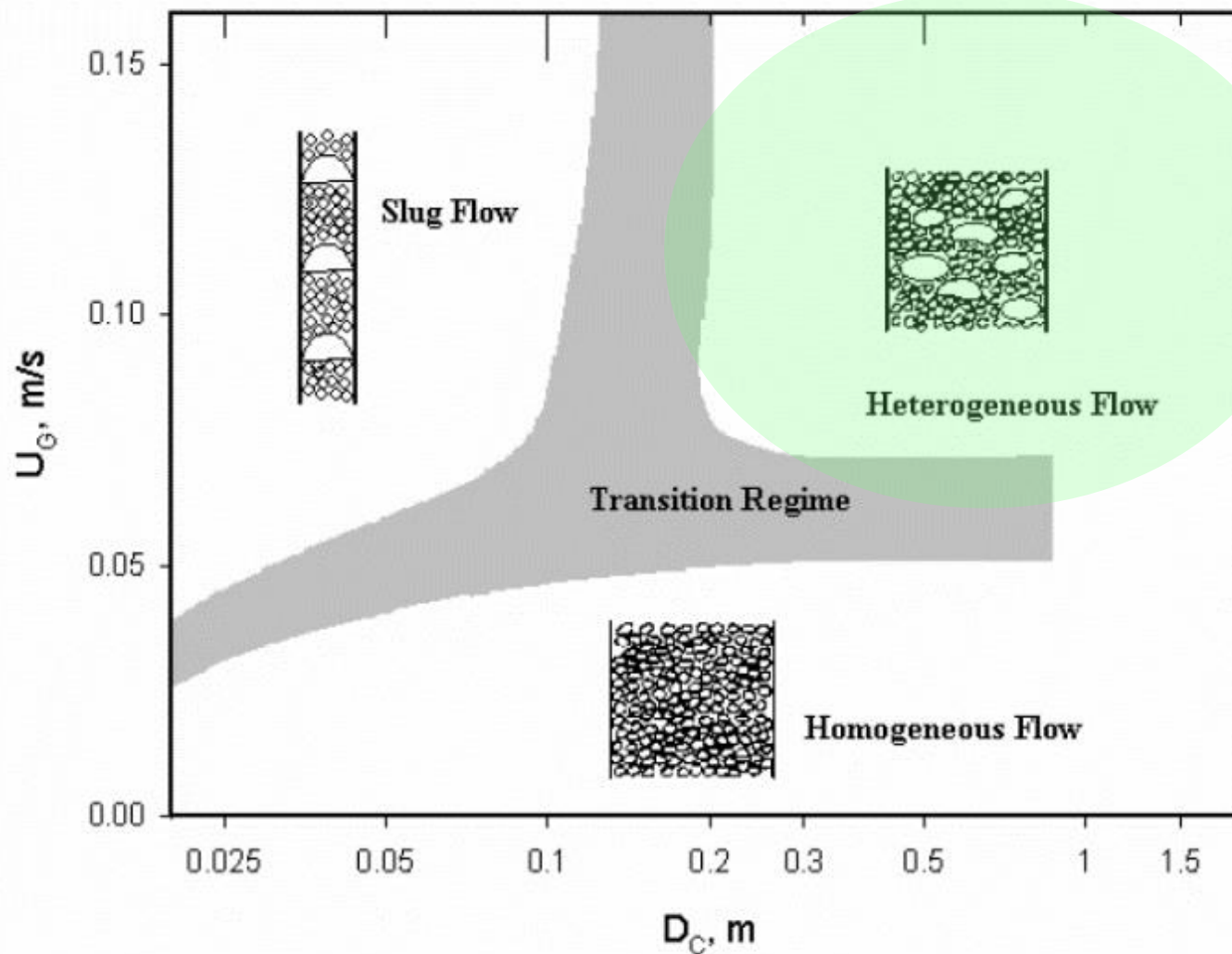
Scaled-up
Plant for 1 bpd



- ✓ Economic optimisation of gas treatment
- ✓ Scaling up of Slurry FT reactor
 - Long term tests of FT synthesis with wood based synthesis gas
 - Upgrading of the raw FT products
 - Testing of FT products



Typical Flow-Regime Map for the Bubble Column Reactor (1)



Upscaling of FT slurry reactor

- Efficiency of bubble column slurry reactor (BCSR) is strongly depended on the hydrodynamic regime of the bubble movement

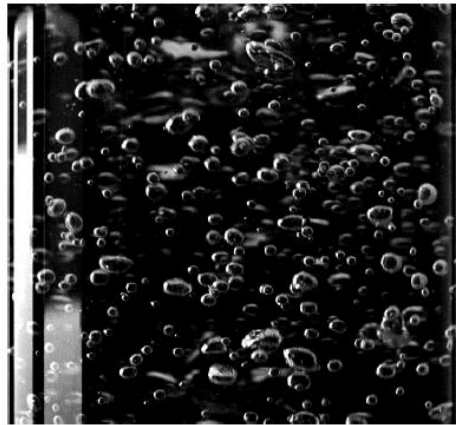


Figure 5: Gas velocity $U_{\text{gas}} = 0,0053 \text{ m/s}$

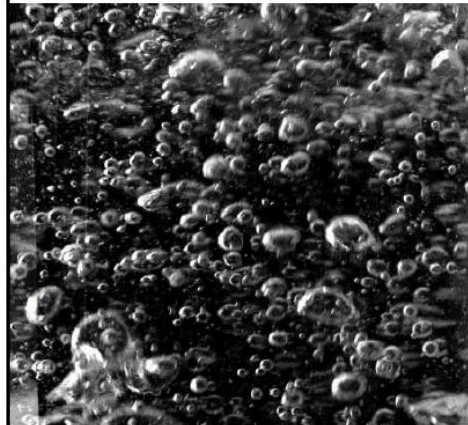


Figure 5: Gas velocity $U_{\text{gas}} = 0,0106 \text{ m/s}$

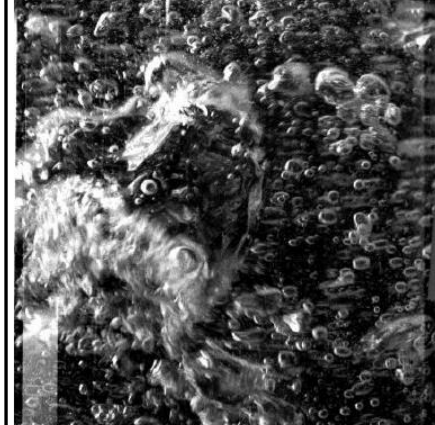
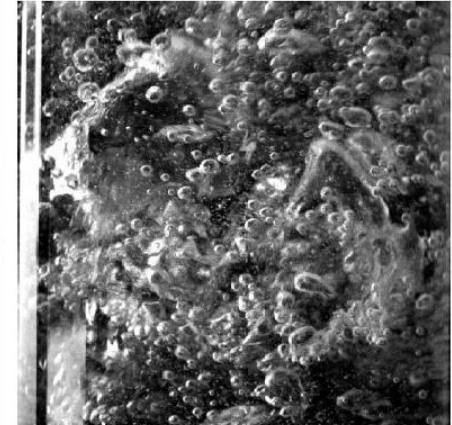


Figure 5: Gas velocities 0,0212 and 0,0371 m/s, respectively



Homogenous Regime

Transition Regime

Heterogeneous Regime

Churn up turbulence increase hold – up of small bubbles → 4 times more efficient than homogenous regime

- Several tests with water and naphtha carried out with cold flow model laboratory plant
 - Variation of gas superficial velocity and slurry concentration
 - Investigation of effect on pressure drop and gas hold – up



- » Switch from water to naphtha is changing hydrodynamics significantly, Weber number (We) can be employed for comparison

$$We = \frac{\rho_{liquid} * U_{gas}^2 * d_{bubble}}{\sigma_{liquid}} \quad (1)$$

Where, in our systems:

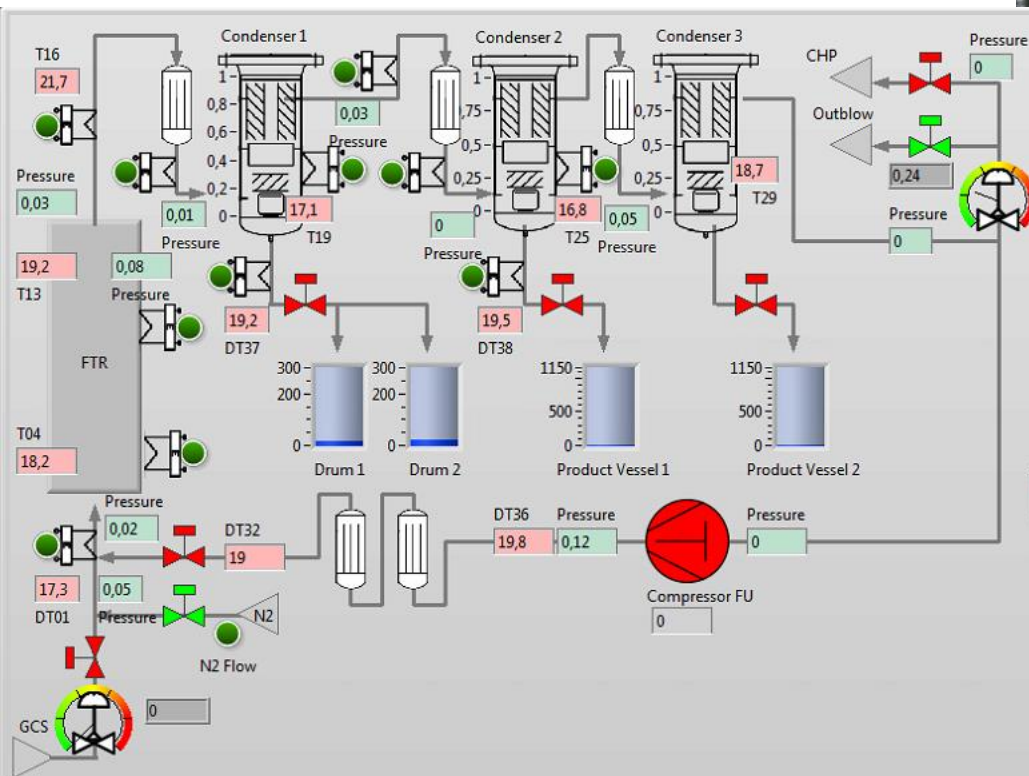
- ρ_{liquid} : density of the liquid;
- U_{gas} : superficial gas velocity through the gas distributor orifices;
- σ_{liquid} : surface tension of the liquid;
- d_{bubble} : initial bubble size at its formation, which can be calculated by [4]:

$$d_{bubble} = \left[\frac{6 * d_{orifice} * \sigma_{liquid}}{g * (\sigma_{liquid} - \sigma_{gas})} \right]^{1/3} \quad (2)$$

With g equals $9,81 \text{ m}^2/\text{s}$.

$We > 2$ bubble breaking and axial mixing => heterogeneous regime!

- Commissioning is ongoing
- First experiments are planned for October 2016



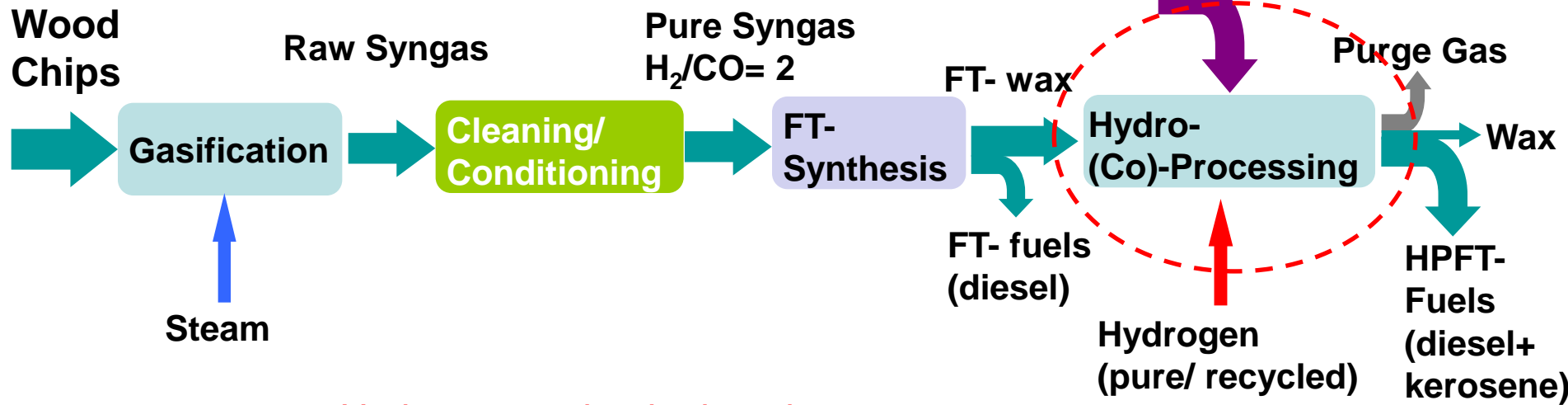
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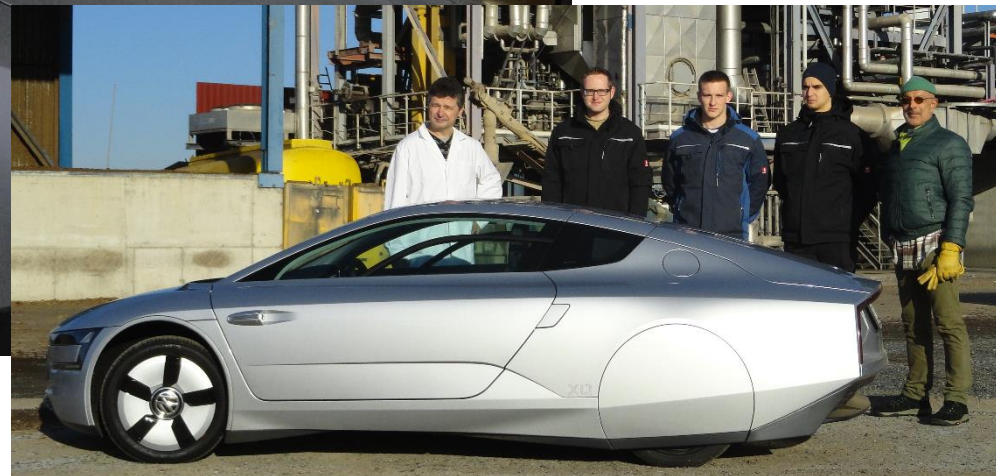
Hydroprocessing is done by partners

Future

Higher efficiency



Energy efficiency
e.g. VW XL1



Selected residues and waste



waste wood A



waste wood B



waste wood C



soft wood pellets



saw dust



PE regrind



SLF-plastics



MSW-plastics



PE+PS



virgin polymers

Information

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<http://www.vt.tuwien.ac.at>

<http://www.bioenergy2020.eu>